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and Applications Study Guide, Chapter 13 In the space at the left, write the letter of the word or phrase that best completes the statement or answers the question.

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13.78 kPa 6. Find the partial pressure of carbon dioxide in a gas mixture with a total pressure of 30.4 kPa if the partial pressures of the other two gases in the mixture are 16.5 kPa and 3.7 kPa. 30.4 kPa 16.5 kPa 3.7 kPa 10.2 kPa Solutions Manual Chemistry: Matter and Change • Chapter 12 237