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CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. c When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released. (d) neither absorbed nor released. 2. d The energy required to remove an electron from a neutral atom is the atom's

5 The Periodic Law

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Chapter 5 Review The Periodic Table Answers

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Chapter 5 Review: Periodic Table. STUDY. PLAY. Family. A group in the periodic table that is categorized as being members of a unit because they have properties in common. Alkali Metals. Soft and Shiny; reacts violently with water; can be cut with a butter knife; never found uncombined with other elements.

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CHAPTER 5 REVIEW The Periodic Law SECTION 5-3 SHORT ANSWER Answer the following questions in the space provided 1 When an electron is added to a neutral atom, energy is (a) always absorbed (c) either absorbed or released (b) always released (d) burned away 2 The energy required

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Science Chapter 5 The Periodic Table. The atomic mass of an element is. Which statement is true about the metal.... At room temperature none of the metals.... Which general statement does NOT apply.... a weighted average of the masses of an element's isotopes. Silicon is a better conductor of electric current than sulfur....

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Modern Chemistry 33 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. ____ In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number.

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Review Chapter 5 The Periodic Table

The idea of arranging the elements in the periodic table according to their chemical and physical properties is attributed to. a. Mendeleev. c. Bohr. b. Moseley. d. Ramsay. ____ 2. Mendeleev left spaces in his periodic table and predicted several elements and their. ... Chemistry Chapter 5 Exam ...

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